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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
$R$ factor $=0.037$
$w R$ factor $=0.083$
Data-to-parameter ratio $=14.0$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## 2-(3-Chlorophenyl)chroman-4-one

The title compound, $\mathrm{C}_{15} \mathrm{H}_{11} \mathrm{ClO}_{2}$, was synthesized from 3-(3-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1-one. The pyrone ring adopts a half-chair conformation.

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## Comment

Flavanones represent important structures of a wide variety of natural products exhibiting various interesting biological activities (Heinisch \& Holzer, 1991 Salvatore et al., 1998). It is also found from biological examination that flavanones, such as 2-(2,3-dimethoxyphenyl)chroman-4-one, which we have reported (Wu et al., 2005), has good molluscicidal activity. Thus, the synthesis of flavanones and their derivatives is of great interest in organic chemistry. We report here the crystal structure of the title compound, (I).

(I)

In (I), the pyrone ring adopts a half-chair conformation (Fig. 1). Atom C9 deviates from the plane defined by atoms C6/C8/C9/C1/O1 by 0.679 (3) $\AA$. The dihedral angle between the $\mathrm{C} 8 / \mathrm{C} 9 / \mathrm{O} 1$ and $\mathrm{C} 1 / \mathrm{C} 6 / \mathrm{C} 7 / \mathrm{C} 8 / \mathrm{O} 1$ planes is $52.13(14)^{\circ}$ and the dihedral angle between the $\mathrm{C} 10-\mathrm{C} 15$ and $\mathrm{C} 1 / \mathrm{C} 6 / \mathrm{C} 7 / \mathrm{C} 8 / \mathrm{O} 1$ planes is $78.72(6)^{\circ}$. The bond lengths, angles and torsion angles in (I) show normal values (Table 1).


Figure 1
The molecular structure of (I), showing $50 \%$ probability displacement ellipsoids and the atom-numbering scheme.


Figure 2
The molecular packing of (I). H atoms have been omitted.

## Experimental

The title compound was synthesized from 3-(3-chlorophenyl)-1-(2-hydroxyphenyl)prop-2-en-1-one ( $4 \mathrm{mmol}, 1.03 \mathrm{~g}$ ) and acetic acid $(0.05 \mathrm{~g})$ in ethanol solution $(12 \mathrm{ml})$ at 353 K over a period of 12 h (yield $43 \%$, m.p. 360-362 K). Crystals of (I) suitable for X-ray diffraction were obtained by slow evaporation of a solution in $95 \%$ ethanol.

## Crystal data

$$
\begin{aligned}
& \mathrm{C}_{15} \mathrm{H}_{11} \mathrm{ClO}_{2} \\
& M_{r}=258.69 \\
& \text { Monoclinic, } P 2_{1} / n \\
& a=11.979(2) \AA \\
& b=5.202(1) \AA \\
& c=19.773(3) \AA \\
& \beta=92.22(2) \AA \\
& V=1231.1(4) \AA^{\circ} \\
& Z=4
\end{aligned}
$$

## Data collection

| Siemens P4 diffractometer | $R_{\text {int }}=0.018$ |
| :--- | :--- |
| $\omega$ scans | $\theta_{\max }=25.5^{\circ}$ |
| Absorption correction: multi-scan | $h=0 \rightarrow 14$ |
| $\quad$ SHELXTL; Sheldrick, 1997) | $k=0 \rightarrow 6$ |
| $T_{\min }=0.853, T_{\max }=0.882$ | $l=-23 \rightarrow 23$ |
| 2790 measured reflections | 3 standard reflections |
| 2296 independent reflections | every 97 reflections |
| 1314 reflections with $I>2 \sigma(I)$ | intensity decay: $1.8 \%$ |

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$w R\left(F^{2}\right)=0.083$
$S=0.87$
2296 reflections
164 parameters
H -atom parameters constrained
$w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0361 P)^{2}\right]$
where $P=\left(F_{o}{ }^{2}+2 F_{c}{ }^{2}\right) / 3$
$(\Delta / \sigma)_{\max }<0.001$
$\Delta \rho_{\text {max }}=0.14 \mathrm{e}^{\AA^{-3}}$
$\Delta \rho_{\text {min }}=-0.20$ e $\AA^{-3}$
Extinction correction: SHELXTL
Extinction coefficient: 0.0140 (14)

Table 1
Selected geometric parameters ( $\left({ }^{\circ},{ }^{\circ}\right)$.

| $\mathrm{Cl}-\mathrm{C} 14$ | $1.745(2)$ | $\mathrm{C} 6-\mathrm{C} 7$ | $1.470(3)$ |
| :--- | :---: | :--- | ---: |
| $\mathrm{O} 1-\mathrm{C} 1$ | $1.366(2)$ | $\mathrm{C} 7-\mathrm{C} 8$ | $1.505(3)$ |
| $\mathrm{O} 1-\mathrm{C} 9$ | $1.450(2)$ | $\mathrm{C} 8-\mathrm{C} 9$ | $1.508(2)$ |
| $\mathrm{C} 1-\mathrm{C} 6$ | $1.402(2)$ |  |  |
|  |  |  | $119.90(18)$ |
| $\mathrm{C} 1-\mathrm{O} 1-\mathrm{C} 9$ | $113.58(14)$ | $\mathrm{C} 1-\mathrm{C} 6-\mathrm{C} 7$ | $114.88(17)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2$ | $117.26(18)$ | $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $111.08(17)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 6$ | $122.39(18)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ |  |
|  |  |  | $175.19(19)$ |
| $\mathrm{C} 9-\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2$ | $-154.59(17)$ | $\mathrm{C} 1-\mathrm{C} 6-\mathrm{C} 7-\mathrm{O} 2$ | $-3.3(3)$ |
| $\mathrm{C} 9-\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 6$ | $24.0(3)$ | $\mathrm{C} 1-\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8$ | $-28.6(2)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | $176.57(18)$ | $\mathrm{C} 6-\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9$ | $-56.2(2)$ |
| $\mathrm{C} 6-\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | $-2.1(3)$ | $\mathrm{C} 1-\mathrm{O} 1-\mathrm{C} 9-\mathrm{C} 8$ | $179.61(17)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 6-\mathrm{C} 5$ | $-175.45(17)$ | $\mathrm{C} 7-\mathrm{C} 8-\mathrm{C} 9-\mathrm{C} 10$ | $-177.41(15)$ |
| $\mathrm{O} 1-\mathrm{C} 1-\mathrm{C} 6-\mathrm{C} 7$ | $6.9(3)$ | $\mathrm{Cl}-\mathrm{C} 14-\mathrm{C} 15-\mathrm{C} 10$ | - |

All H atoms were placed in calculated positions and treated using a riding model, with $\mathrm{C}-\mathrm{H}$ distances of $0.93-0.98 \AA$ and $U_{\text {iso }}(\mathrm{H})$ values of $1.2 U_{\text {eq }}(\mathrm{C})$.

Data collection: XSCANS (Siemens, 1994); cell refinement: XSCANS; data reduction: SHELXTL (Sheldrick, 1997); program(s) used to solve structure: $S H E L X T L$; program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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